TRAINING OF TRAINERS PROGRAMME ON CAPACITY DEVELOPMENT OF ETP OPERATORS

Promotion of Sustainability in the Textile and Garment Industry in Asia - FABRIC







Day 4: Presentation 4

Tertiary Treatment for Textile Wastewater



Contents

Objectives of tertiary treatment

Disinfection as a tertiary treatment

Filters for polishing treated effluent

Chemical oxidation for organic removal from treated effluent

- Final treatment stage, mostly to comply with norms
- Focus on
 - ✓ Reduction of color
 - ✓ Reduction of suspended solids
 - ✓ Destruction of pathogens
 - ✓ Removal of organics
 - ✓ Improvement of treated effluent appearance
 - ✓ sometimes for aesthetic purpose and as precautionary or complimentary measure

- Required as **pre-treatment for effluent recovery** using membrane systems by removing turbidity, hardness etc.
- Single stage or using combination of tertiary systems.
- Often installed as polishing treatment after physicochemical treatment, in most primary ETPs and referred to as tertiary treatment



Common tertiary treatment systems

- **Disinfection** mainly to kill micro-organisms in treated effluent and some for organic removal.
- Filters, using filter media to filter out suspended particles in effluent
- Adsorption filters most commonly activated carbon filters to remove organics
- Oxidation systems to oxidize residual organics in treated effluent

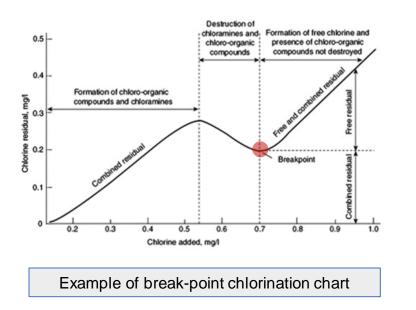


Other tertiary treatment systems

- Chemical precipitation systems for removal of phosphates/metals.
- Softening using lime/soda softening or zeolite softeners
- Membrane based filtrations (using ultra filters or nanofilters)

Disinfection

- To kill micro-organisms, specifically pathogens in treated effluent
- Chlorination most common disinfection system
 - mixing effluent with chlorine gas in contact chambers or dosage of hypo-chlorites
 - ✓ Chlorine killing micro-organisms by breaking their cell walls
 - ✓ In case of sodium or calcium hypo-chlorites, chlorine content to be calculated and dosage fixed accordingly
- Generation of disinfection-by-products (DBPs)
 - \checkmark haloacetic acid, trihalomethane, and chloral hydrate
 - \checkmark controlled by activated carbon filtration or membrane filters.



Disinfection using chlorination

- Chlorination usually based on break-point chlorination
 - Keep adding chlorine (or hypo) to measured quantity of effluent
 - Check residual chlorine using DPD laboratory tablets
 - ✓ Residual chlorine first increasing, then decreasing and increasing again with more chlorine dosing
 - Point of increase = break point or correct dosage of chlorination.

Disinfection using chlorination

- Storage and dosing chlorine gas difficult and safety risk
 - Use of sodium hypochlorite or calcium hypochlorite in small- and medium ETPs but less preferred due to sludge issues
 - ✓ Sodium hypochlorite generally containing 10% 12% chlorine
 - Need to calculate dosage accordingly
- Increased efficiency of chlorination with higher dosage, lower pH, higher temperature and longer contact time (usually 30 min)
 - If not effective, take corrective actions such as by increasing dosage, increasing contact time (reduce flow) or reducing pH

Disinfection using Ultraviolet (UV) system

- Pathogens killed by exposing effluent to UV radiation damaging DNA of bacteria/virus
 - effluent passing through chamber illuminated by UV rays from UV lamp
 - \checkmark low pressure and medium pressure lamps common.
 - Medium handling higher flows, but consuming more power
- As per wavelength classification into UV-A, UV-B, UV-C
 - \checkmark UVA less powerful, but consuming less power
 - ✓ UV-B with medium efficiency and medium power consumption
 - \checkmark UV-C highest power consumption and efficiency

UV disinfection

- Effective in bacteria, viruses and cysts
- No residual effect
- No need to buy, store dangerous chemicals.
- Short contact time (20-30 sec)
- Less space requirement

- Low dosage not effective.
- Organisms sometimes surviving
- Frequent cleaning
- Not suitable for TSS levels above 30 mg/l
- Costlier in installation

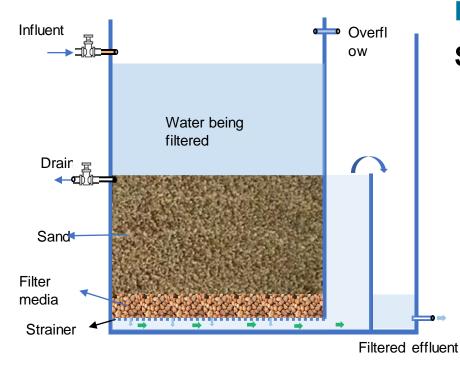
Rapid sandfilter Advantages

Rapid sandfilter disadvantages

Image: AOS treatment solutions

Filtration

- Used for removal of suspended solids in treated effluent
 - Also partly reducing BOD/COD by removing some organics (like MLSS particle) in the suspended solids
- Done by gravity for pressure filters
 - Slow sand filters using gravity (similar to sludge drying beds)
 - Pressure sand filters using vessel filled with filter media with effluent being pumped and filter under pressure
 - ✓ Fine filtrations (such as pre-treatment of membrane) with cartridge filters



Filtration - Gravity sand filters

Slow sand filters

- similar to sludge drying beds with coarse media at bottom, fine sand at top
- water admitted from top, with pressure by water column speeding filtration
- · Solids retained in top sand layer
- Periodically, filter dried and solids scooped out for disposal
- Top sand cleaned and topped up with fresh sand

Slow sand filter

- Simple construction
- very low operating costs
- simple process control
- good efficiency

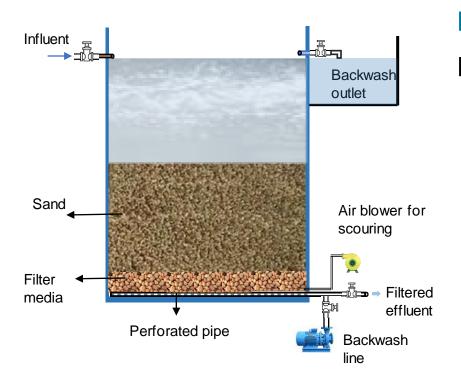
- high land requirement
- suitable for small ETP
- potential clogging
- not suitable for effluent with high level of suspended solids

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Slow sandfilter Advantages



Slow sand filter disadvantages



Filtration - Gravity sand filters

Rapid sand filters

- **similar to slow sand filter**, provision in filtrate line to admit water at pressure to carry out periodical backwash
- network of net covered perforated pipes for draining & backwashing.
- backwashing with pump or water tank at sufficient height for required head
- Often with air scouring as additional washing aid

Rapid sand filter

- higher capacity
- no manual cleaning required
- suitable for medium ETPs
- easy process control
- good efficiency

- Land requirement (less than slow sand filter, higher than pressure filters)
- not suitable for effluent with high suspended solids
- relatively more maintenance

Rapid sandfilter Advantages



Rapid sandfilter disadvantages



Filtration – Pressure filters

- Common types in wastewater treatment:
 - ✓ Pressure sand filters
 - ✓ Multi-grade filters
 - ✓ **Dual media** filters
- Similar in construction and operational pattern but varying in composition of filter media
- made of FRP, MS (often rubber lined) and stainless steel
- New types of media coming to market every year.

Pressure filters - different vessel materials



Mild Steel (MS)



Fibre reinforced Plastic (FRP)



Stainless Steel (SS)



Filtration – Pressure filters

- Measurement of operational efficiency
 - ✓ **filtration rate** = quantity of water filtered per unit area
 - head loss = difference between inlet and outlet pressure
 - ✓ frequency of backwashing needed.
- Backwashing with clean water whenever pressure drop across filter more than 1 bar.
- Sometimes preceded by air scouring for agitating media with scrubbing action and loosens retained solids



Filtration – Pressure filters

- Common filtration media
 - $\checkmark\,$ most common silica sand and anthracite coal
 - \checkmark quartz sand, garnet, magnetite
- Size and shape of filter media affecting efficiency
 - Smooth and rounded better than sharp and angular media
 - Most suspended solids at surface (top 5 10 cms), gradually solids percolating down to prevent rapid pressure drop

Pressure filters: Operation cycle

Service

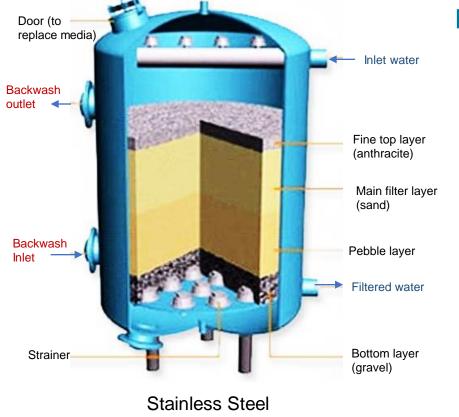
- Inlet water is pumped down through the media via the distribution tube.
- Collect the drained water as filtered water.

Backwash

- Flow is reversed.
- The flow is forced through the bottom and up through the media.
- Backwash lifts media and causes scouring
- Collected dust and debris is flushed to the drain

Slow Rinse

- Use clean water allowed to flow down through media bed & distribution tube to drain.
- With autovalves and controls, entire backwash & rise can be programmed based on fixed quantity of flow, at scheduled time or based on differential pressure.



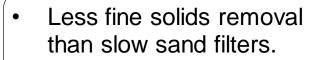
Filtration: Pressure sand filters

- Usually cylindrical vessel filled with filter media.
- Vertical or horizontal orientation
- Set of frontal pipe work and valves
- Graded silica quartz sand
- Sand layer supported by under-bed of pebble/gravel.
- Water admitted via top distributor
- Under-drain collecting filtered water

Pressure sand filter

- High flow rate
- no manual cleaning needed
- suitable for medium/large ETPs
- easy process control
- Low land requirement

Pressure sand filter Advantages



- Need energy for its operation.
- Backwash water requirement high.

Pressure sand filter disadvantages

26 04-09-2023 ToT-ETP Operators



Filtration : Multigrade filters

- similar to pressure sand filter in construction with cylindrical vessel and identical piping/valves
- Same way of operation and backwashing
- coarse and fine media mixed together in fixed proportion.
- filtration efficiency not as fine as in pressure sand filter but turbidity reduction better

Multi grade filter

- Substantially higher specific flow rate than pressure sand filter
- Turbidity reduction better than PSF
- no manual cleaning needed
- Low land requirement

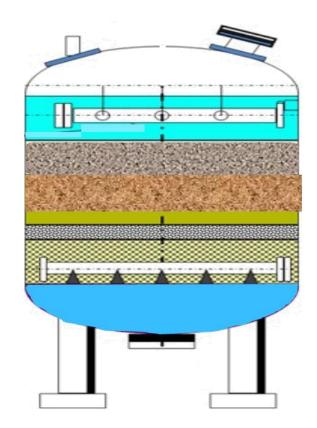
Pressure sand filter Advantages



 filtration efficiency not as fine as in pressure sand filter

- Need more energy for its operation.
- Backwash water requirement high.

Pressure sand filter disadvantages



Filtration – Dual media filters

- similar to pressure sand filter in construction with cylindrical vessel and identical piping/valves
- Same way of operation and backwashing
- sand-anthracite filter or multi-media used for removal of turbidity and suspended solids
- Remove TSS as low as 10 20 microns

Dual media filter

- Very efficient particle removal
- high filtration rate
- higher specific flow rate than PSF & MGF
- number of filters and size for ETP still smaller

Pressure sand filter Advantages



 Backwashing frequency needed for DMF is higher than PSF and MGF

- Backwash water consumption is much higher.
- Life of media is lesser.

Pressure sand filter disadvantages



Pressure Filters

Maintenance requirements

- If made in mild steel, periodical painting with epoxy coating needed
- Once a week:
 - Check of all valves, flanges and gaskets for its tightness.
 - Check for any leaks => to be arrested promptly
 - Check of pressure gauges, auto valves for their correct operation.



Pressure Filters

Maintenance requirements

- Irrespective of media, media degradation over period of time
 - \checkmark more predominant with natural media
 - ✓ salt in effluent, pH variations etc. chemically degrading media
 - ✓ abrasion by flowing water physically degrading media
- Need to replenish or replace after period of time
 - ✓ Media removed through bottom door
 - ✓ Refilled through trap door at top



(2) Filtration – Pre-coat filters

- filters or flexible screens on which coat of filter medium given
- temporary or fixed to mechanical screen
- Filter media
 - ✓ inert materials of fine fibrous or granular structure e.g. diatomaceous earth (diatomite).
 - ✓ Other media: Perlite, powered organic rock, activated carbon, asbestos and cellulose



Filtration – Cartridge filters

- used for very fine filtration e.g. pre-treatment of membranes
- cartridge filters considered as consumables
 - \checkmark possible to clean by soaking in cleaning solution
 - \checkmark to be replaced once clogged irreversibly
- usually very small in construction
- generally used in-line of pumping lines
- usually pore sizes in range of 0.2 20 microns
 - ✓ smaller pore size = shorter replacement period
- There are surface filters & depth filters

To remember



- Filters common in tertiary or polishing treatment
- High efficiency of suspended solids and turbidity removal and easiness of control advantages of filtration vis-à-vis other tertiary treatment options
- Filters **susceptible to clogging** by suspended solids and not suitable if high TSS levels
- Recent developments in design of filters using light
 weight media with high uniformity coefficients
- Pressure filters ideal for tertiary treatment units in Bangladesh because of low space requirements

Image: Center Enamel

Basic concept and overview of tertiary treatment

Options for management of residual organics

- Treated effluent containing organics not removed in biological treatment since on bio-degradable to lesser extent
- Tertiary treatment systems for removal of residual organics:
 - ✓ Adsorption of organics in adsorbent media, such as activated carbon filters and organic scavengers.
 - ✓ Advanced oxidation systems
 - ✓ Ozonation of treated effluent
 - ✓ Fenton treatment: Oxidation catalyzed by iron

Basic concept

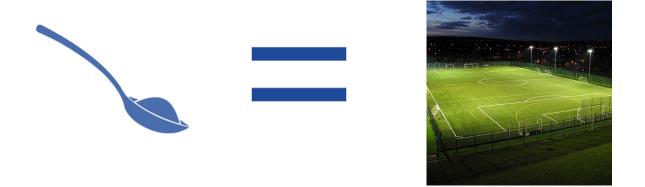
Adsorption = adhesion of ions or molecules to surface

- In effluent treatment entrapment of organics (or other contaminants like chlorine) in adsorbent medium
- Physical entrapment in voids of porous medium or attachment to surface due to surface charge
- Activated carbon: Inert solid adsorbent material made from almost any carbon containing feedstock (e.g. wood, coconut shells and coal)
- Porous, inexpensive and high surface area per gram

Basic concept

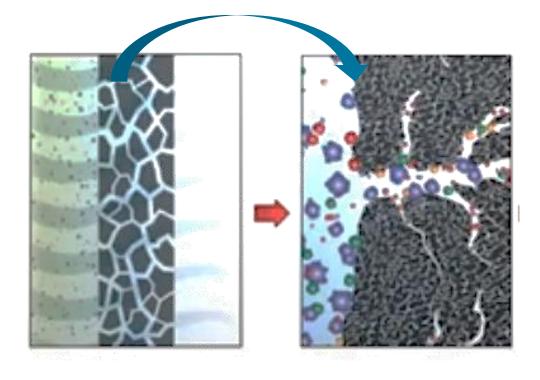
Activated carbon

• One teaspoon of activated carbon more surface area than one football field!



Activated carbon filters

- similar in construction to pressure sand filters
- activated carbon as filter media
- commonly granulated activated carbon with 0.4 1 mm diameter or powdered activated carbon
- backwash process similar to pressure sand filters but without air scouring
- strainers at bottom to prevent carbon from flowing out with filtered and at top to prevent loss of carbon during backwash
- carbon media to be replaced once exhausted
- regeneration presently not economical



Activated carbon filters

Concept

- Organic molecules trapped in pores of carbon media
- Subsequent organic inflow pushing trapped material into micropores
- Process continuing till media fully exhausted

Activated carbon filters

General specification for activated carbon suitable to textile effluent

Parameter	Value needed
Min Moisture, percent by mass (max)	5
Ash, percent by mass (max)	2
Hardness number, Min	90
Min Adsorption capacity- iodine number	450
Half dichlorination value, cm (max)	7
Surface area, m2/g (min)	550

Advantages of Activated carbon filter



DIsadvantages of Activated carbon filter



Image: NCH Asia

Advanced oxidation processes = chemical treatment process for removing organic (and sometimes inorganic) pollutants

- Common systems
 - ✓ Ozone
 - ✓ **Hydrogen peroxide** with or without UV radiation
 - ✓ Fenton treatment
- Concept
 - \checkmark Hydroxyl radical (OH-) and nascent oxygen as active reactants
 - ✓ Hydroxyl radicals produced in water with primary oxidants like oxygen, ozone and peroxides enhanced with energy sources or catalysts

Photochemical oxidation process

- Hydroxyl radicals present in chemicals with extra oxygen atoms
 ✓ generation enhanced by radiation with UV rays
 > H2O2 + UV → 2·OH
- Organics pollutants oxidized by hydroxyl radical and broken into simpler organics and further oxidized into carbon dioxide
 - ✓ **Higher efficiency in acidic condit**ions (optimal pH 3 6)
 - Natural organic matter or carbonate species reducing effectiveness
 - ✓ Reduced metal ions (e.g. Ferrous and Manganous) reducing effectiveness since consuming excess oxygen

Example views of advanced oxidation systems



AOP system by NOVEXX



AOP system by Enviro Chemie

By-product management

- Concerns about toxic by-products despite oxidization and neutralization of toxic and hazardous organics present in textile effluent
 - Possibility of highly toxic by-products from partial degradation of dissolved organic
 - ✓ Bromate and excess peroxide
 - ✓ If chlorine used, halogenated organic by-products. e.g. toxic chlorophenols.
- By-products depending on composition of effluent (!)
 - ✓ Consider of advanced oxidation based on prior analysis of treated effluent and analysis of effluent from advanced oxidation processes

Advantages

- Low space requirement even for high capacity units
- **Complete degradation of organics** into water, carbon dioxide, and salts (Mineralization)
- Fast reaction and very lower retention times compared to conventional treatment processes
- Treatment of wide range of organics (all organic materials, some heavy metals)
- **Complete disinfection** besides organic degradation
- No sludge production

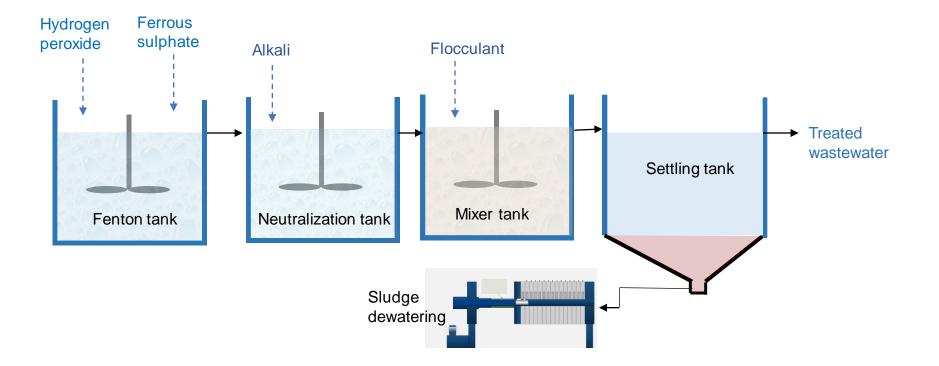
Disadvantages

- Need for highly skilled labor to operate and control
- High capital and operating & maintenance costs (energy, chemical reagents)
- **Complex** chemistry tailored to specific contaminants
- Good understanding required for selection of technology because of several different variants
- Need to control and remove of residual peroxide, if hydrogen peroxide based system used
- Residuals affecting membranes if proper anti-oxidant control not ensured

Fenton treatment

- Popular variant type of advanced oxidation (also as further advanced photo Fenton system.
- Based on liberation of OH radicals from H₂O₂ catalyzed by ferrous ions (in photo Fenton with use of UV radiation)
 - oxides of iron produced in reaction catalyzing oxidation of organics by OH radicals

Fenton treatment



Fenton treatment



Fenton treatment unit by Xh2o Solutions Pvt. Ltd

Fenton treatment with modified Fenton reactors



Fluidized Fenton reactor (Source: Science Direct)



Photo Fenton reactor (Model: ENVIOLET)

Fenton Treatment

- Relatively lower capital cost
- Simple and easy process
- Suitable for all organic materials and some heavy metals
- No concentration of contaminants like salts

- Generation of ferric sludge for dewatering/disposal
- High operation and maintenance costs: peroxide and pH management
- Need to adhere to strict pH range

Fenton system Advantages



Fenton system disadvantages

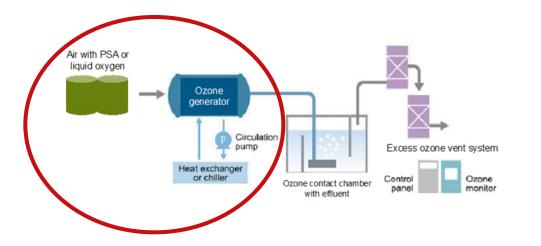
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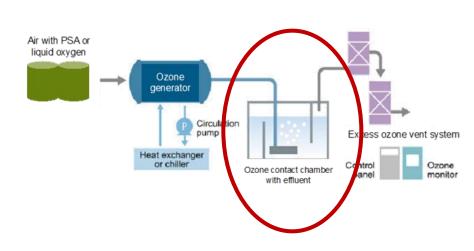
Basic concept

- Ozone (O3) = oxygen gas with additional oxygen atom
 - pale blue gas with distinctively pungent smell and potentially toxic
- Ozone generated in most ETPs from oxygen-bearing gas subjected to electric field or UV
 - done on-site since unstable and quickly decomposing to oxygen
 - ✓ Ozone generators using air or oxygen as source, with occasional oxygen concentrators
- when generated from air usual concentration 0.5-2% ozone
- with oxygen gas usual concentration 4 7% ozone

Use in effluent treatment

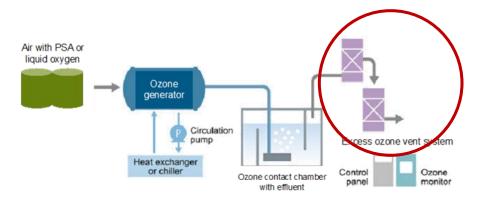
- Electrical discharge method most common source for generating ozone
 - ✓ Extremely dry air or pure oxygen exposed to controlled, uniform high-voltage discharge





Use in effluent treatment

- After generation, ozone fed into down-flow contact chamber containing wastewater to be treated
 - ✓ Aim to transfer ozone from gas bubble into bulk liquid with sufficient contact time for disinfection.
 - ✓ Commonly used contactor type diffused bubble
- co-current and counter-current
- variants: positive pressure injection, ventury, mechanically agitated and packed tower.



Use in effluent treatment

- Treatment of off-gases from to destroy any remaining ozone
 - ✓ In case of pure oxygen as feed-gas, recycling of off-gases from contact chamber possible to generate ozone or reuse in aeration tank.

Example



Ozonator (at Wylie Water Treatment Plant, North Texas)

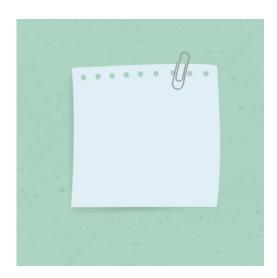
Advantages

- Little space required
- Very effective in destroying pathogens and residual organics
- Short treatment time of less than 30 min
- No harmful residuals since ozone decomposing rapidly
- No bacteria regrowth
- Onsite generation of ozone avoiding safety issues with shipping and handling
- Increase in dissolved oxygen (DO) concentration of effluent eliminating need for reaeration positively affecting DO in receiving stream

Disadvantages

- High capital and operation & maintenance costs (high power consumption)
- Need for highly skilled labor to operate and control
- Not very effective at low concentration
- More complex than other tertiary units requiring complicated equipment and efficient contacting systems
- Need for corrosion-resistant material (e.g. stainless steel)
- Not economical for removal of high levels of TSS/COD
- Very toxic nature of ozone and off-gases

To remember



- Adsorption and oxidation common options for removing organics
- Activated carbon treatment quite common in Bangladesh textile ETPs as polishing treatment
 - ✓ **Need to replenish carbon** after media exhausted (!)
- Fenton treatment installed in few ETPs
 - ✓ internationally preferred advanced oxidation method
 - ✓ **No sludge** generation and **low space** requirement
- Advanced oxidation technologies
 - costlier than other tertiary systems
 - ✓ suitable for effluent with **low residual organics**

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