



Tertiary treatment – Part 2

GIZ FABRIC – ETP Operator Course



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Basic concept and overview of tertiary treatment

Basic concept and overview of tertiary treatment

Options for management of residual organics

- Treated effluent containing organics not removed in biological treatment since on bio-degradable to lesser extent
- Tertiary treatment systems for removal of residual organics:
 - Adsorption of organics in adsorbent media, such as activated carbon filters and organic scavengers.
 - Advanced oxidation systems
 - Ozonation of treated effluent
 - Fenton treatment: Oxidation catalyzed by iron

Basic concept

Adsorption = adhesion of ions or molecules to surface

- In effluent treatment **entrapment of organics** (or other contaminants like chlorine) in adsorbent medium
 - Physical entrapment in voids of porous medium or attachment to surface due to surface charge
 - Activated carbon: Inert solid adsorbent material made from almost any carbon containing feedstock (e.g. wood, coconut shells and coal)
 - Porous, inexpensive and high surface area per gram
- Do you know?..a teaspoon of activated carbon has more surface area than a football field!

Basic concept

Activated carbon

One teaspoon of activated carbon more surface area than one football

field!





Basic concept

Activated carbon

Produced by

- **Physical activation**
- **Activation/Oxidation**
- Chemical activation



Basic concept

Activated carbon

- Physical activation
 - Carbonization with carbon containing material being pyrolyzed at temperatures in range 600–900 °C
 - usually in an inert atmosphere with gases like argon or nitrogen
- Activation/Oxidation
- Chemical activation



Basic concept

Activated carbon

- Physical activation
- **Activation/Oxidation**
 - Raw material exposed to oxidizing atmospheres (oxygen or steam) at temperatures above 250 °C, usually in range of 600-1200 °C.
- Chemical activation



Basic concept

Activated carbon

- Physical activation
- Activation/Oxidation
- Chemical activation
 - Carbon material impregnated with chemicals (e.g acid, alkali, salt)
 - subjected to higher temperatures (250 600°C).
 - temperature activating carbon at this stage by forcing material to open and have more microscopic pores
 - Preferred activation option



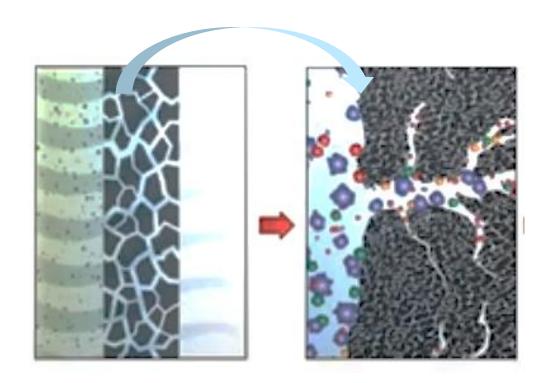
Activated carbon filters

- similar in construction to pressure sand filters (see presentation 11.1)
- activated carbon as filter media
 - commonly granulated activated carbon with 0.4 1 mm diameter or powdered activated carbon
- backwash process similar to pressure sand filters but without air scouring
- strainers at bottom to prevent carbon from flowing out with filtered and at top to prevent loss of carbon during backwash
- carbon media to be replaced once exhausted
 - regeneration presently not economical

Activated carbon filters

Concept

- Organic molecules trapped in pores of carbon media
- Subsequent organic inflow pushing trapped material into micropores
- Process continuing till media fully exhausted



Activated carbon filters

General specification for activated carbon suitable to textile effluent

Parameter	Value needed
Min Moisture, percent by mass (max)	5
Ash, percent by mass (max)	2
Hardness number, Min	90
Min Adsorption capacity- iodine number	450
Half dichlorination value, cm (max)	7
Surface area, m ² /g (min)	550

Activated carbon filters

Operation and backwashing

- During regular operation...
 - pump inlet water into filter
 - observe filtrate for its clarity



Activated carbon filters

Operation and backwashing

- For backwashing...
 - take filter off line by switching off feed line and close feed valve
 - open backwash line and backwash drain valves
 - pass clean backwash water upwards through filter bed and allow to drain
 - observe clarity of drained water and ensure no carbon drained out
 - once carbon powder coming out, first indication of media's lifetime end
 - close drain valve and backwash valves and open feed/filtrate lines

Activated carbon filters

Maintenance

- If filters made of mild steel, periodical painting (epoxy coating) needed
- Weekly checks of
 - all valves, flanges and gaskets for its tightness.
 - for any leaks => to be arrested promptly
 - pressure gauges, auto valves for correct operation.

Activated carbon filters

Maintenance

- Replacement of media
 - Since adsorption based on available voids in activated carbon, voids getting filled up and stopping further organic removal
 - Indicated by increased color and turbidity in filtered water
 - Removal of exhausted media through trapdoor at top
 - Cleaning of filter inside and flushing pipelines
 - Check and, if necessary replacement of strainers
 - Filling of filter with fresh granular media

Activated carbon filters

Advantages

- High efficiency removing color and residual organics
- Simple and easy to control and maintain
- Low residuals without sludge => exhausted carbon only residual
- Filter made of natural materials like bituminous, wood, coconut shell
- Ability to remove difficult contaminants e.g. complex organics or metal salts



Activated carbon filters

Disadvantages

- No removal of dissolved salts
- Limited life of media requiring replacement
 - Fast exhaustion with high pollutant load or high chlorine levels at inlet
- Limited effectiveness against bacteria, virus, pathogens
- Disposal of exhausted media of concern
- High operating cost depending on pollutant type and load
 - In general inexpensive treatment (!)



Advanced oxidation processes = **chemical treatment** process for **removing organic** (and sometimes inorganic) **pollutants**

- Common systems
 - Ozone
 - Hydrogen peroxide with or without UV radiation
 - Fenton treatment
- Concept
 - Hydroxyl radical (OH-) and nascent oxygen as active reactants
 - Hydroxyl radicals produced in water with primary oxidants like oxygen, ozone and peroxides enhanced with energy sources or catalysts

Photochemical oxidation process

- Hydroxyl radicals present in chemicals with extra oxygen atoms
 - generation enhanced by radiation with UV rays
 - $H2O2 + UV \rightarrow 2 \cdot OH$
- Organics pollutants oxidized by hydroxyl radical and broken into simpler organics and further oxidized into carbon dioxide
 - Higher efficiency in acidic conditions (optimal pH 3-6)
 - Natural organic matter or carbonate species reducing effectiveness
 - Reduced metal ions (e.g. Ferrous and Manganous) reducing effectiveness since consuming excess oxygen

Example views of advanced oxidation systems



AOP system by NOVEXX



AOP system by Enviro Chemie



By-product management

- Concerns about toxic by-products despite oxidization and neutralization of toxic and hazardous organics present in textile effluent
 - Possibility of highly toxic by-products from partial degradation of dissolved organic
 - Bromate and excess peroxide
 - If chlorine used, halogenated organic by-products. e.g. toxic chlorophenols.
- By-products depending on composition of effluent (!)
 - Consider of advanced oxidation based on prior analysis of treated effluent and analysis of effluent from advanced oxidation processes

Advantages

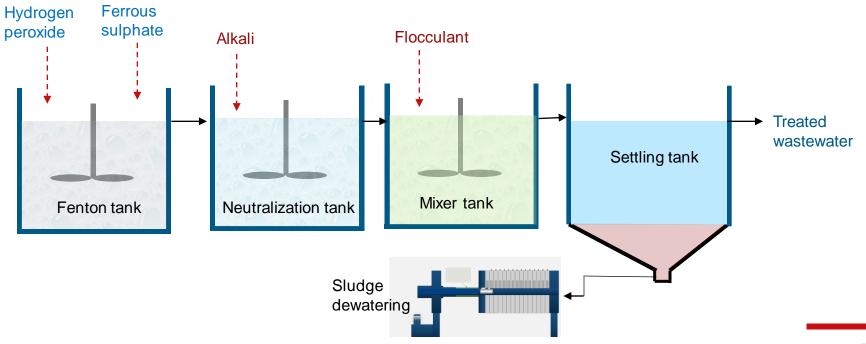
- Low space requirement even for high capacity units
- Complete degradation of organics into water, carbon dioxide, and salts (Mineralization)
- Fast reaction and very lower retention times compared to conventional treatment processes
- Treatment of wide range of organics (all organic materials, some heavy metals)
- Complete disinfection besides organic degradation
- No sludge production

Disadvantages

- Need for highly skilled labor to operate and control
- High capital and operating & maintenance costs (energy, chemical reagents)
- Complex chemistry tailored to specific contaminants
 - Good understanding required for selection of technology because of several different variants
- Need to control and remove of residual peroxide, if hydrogen peroxide based system used
- Residuals affecting membranes if proper anti-oxidant control not ensured

- Popular variant type of advanced oxidation (also as further advanced photo Fenton system.
- Based on liberation of OH radicals from H₂O₂ catalyzed by ferrous ions (in photo Fenton with use of UV radiation)
 - oxides of iron produced in reaction catalyzing oxidation of organics by OH radicals

- Optimum pH 3 5
 - At higher pH, less effective because of iron precipitation as ferric hydroxide
 - At lower pH, OH radicals using excess H+ ions and affecting treatment
 - Note: pH of mixture dropping during reaction process by partly prevented by using ferrous sulphate and adding H_2O_2 in stages.





Fenton treament unit by Xh2o Solutions Pvt. Ltd

Fenton treatment with modified Fenton reactors



Fluidized Fenton reactor (Source: Science Direct)



Photo Fenton reactor (Model: ENVIOLET)

Fenton treatment

Advantages

- Relatively lower capital cost compared to other advanced oxidation processes
- Simple and easy process control with less automation needed
- High efficiency with high level of organics degradation into carbon dioxide
- Suitable for all organic materials and some heavy metals
- No concentration of contaminants like salts

Fenton treatment

Disadvantages

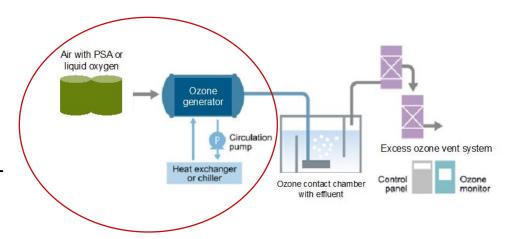
- Generation of ferric sludge to be dewatered and disposed off
- High operation and maintenance costs due to peroxide and pH management
- Need to adhere to **strict pH range**

Basic concept

- Ozone (O3) = oxygen gas with additional oxygen atom
 - pale blue gas with distinctively pungent smell and potentially toxic
- Ozone generated in most ETPs from oxygen-bearing gas subjected to electric field or UV
 - done on-site since unstable and quickly decomposing to oxygen
 - Ozone generators using air or oxygen as source, with occasional oxygen concentrators
 - when generated from air usual concentration 0.5-2% ozone
 - with oxygen gas usual concentration 4 7% ozone

Use in effluent treatment

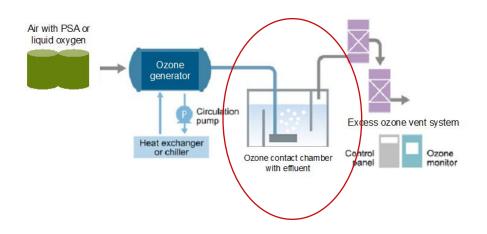
- Electrical discharge method most common source for generating ozone
 - Extremely dry air or pure oxygen exposed to controlled, uniform highvoltage discharge



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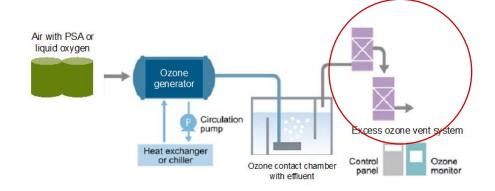
Use in effluent treatment

- After generation, ozone fed into down-flow contact chamber containing wastewater to be treated
 - Aim to transfer ozone from gas bubble into bulk liquid with sufficient contact time for disinfection.
 - Commonly used contactor type diffused bubble
 - co-current and counter-current
 - variants: positive pressure injection, ventury, mechanically agitated and packed tower.



Use in effluent treatment

- Treatment of off-gases from to destroy any remaining ozone
 - In case of pure oxygen as feed-gas, recycling of off-gases from contact chamber possible to generate ozone or reuse in aeration tank.



Example





Ozonator

(at Wylie Water Treatment Plant, North Texas)

Advantages

- Little space required
- Very effective in destroying pathogens and residual organics
- Short treatment time of less than 30 min
- No harmful residuals since ozone decomposing rapidly
- No bacteria regrowth
- Onsite generation of ozone avoiding safety issues with shipping and handling
- Increase in dissolved oxygen (DO) concentration of effluent eliminating need for reaeration positively affecting DO in receiving stream

Disadvantages

- High capital and operation & maintenance costs (high power consumption)
- Need for highly skilled labor to operate and control
- Not very effective at low concentration
- More complex than other tertiary units requiring complicated equipment and efficient contacting systems
- Need for corrosion-resistant material (e.g. stainless steel)
- Not economical for removal of high levels of TSS/COD
- Very toxic nature of ozone and off-gases

To remember

- Adsorption and oxidation common options for removing organics
- Activated carbon treatment quite common in Bangladesh textile ETPs as polishing treatment
 - Need to replenish carbon after media exhausted (!)
- Fenton treatment installed in few ETPs
 - internationally preferred advanced oxidation method
 - No sludge generation and low space requirement
- Advanced oxidation technologies
 - costlier than other tertiary systems
 - suitable for effluent with low residual organics



